Notes

- Next week:
 - Guest lecture on Tuesday
 - Your project presentations on Thursday
 - Make sure you can produce animations for then (if you have problems, let me know ASAP)
 - If you're not using MPEG or QuickTime, send me a sample animation in your chosen format to make sure I can display it in class!
- Teaching evaluations today

Clouds

- · Another extension of incompressible flow
- Basic physics:
 - Humid air rises (due to buoyancy)
 - As altitude increases, temperature and pressure decrease
 - Sooner or later (started by dust or other particles)
 water has to condense
 - Very tiny water droplets form (radius about 10μm), hundreds per cubic cm --- this is the visible cloud
 - (and rain, snow, hail etc. form when droplets get bigger, start to merge, freeze, etc.)
 - Condensation releases heat, causes more buoyancy

Basis

- [Harris '03] "Simulation of cloud dynamics on graphics hardware"
- Need to track
 - Velocity u
 - Potential temperature $\boldsymbol{\theta}$
 - Mass fraction of water vapour ${\boldsymbol{q}}_{v}$
 - Mass fraction of condensed water $\ensuremath{\mathsf{q}}_{\ensuremath{\mathsf{c}}}$
- The new variables feed into buoyancy force for velocity
- Advection + thermodynamics lets us solve for the new variables

Heat and temperature

- Heat (energy) is fundamental
 - Absolute temperature also depends on pressure, heat capacities, etc.
- When background conditions are basically constant, heat and temperature are proportional
- In the atmosphere, they're not
- So we should use something more directly related to heat than absolute temperature

Potential Temperature

- Density, pressure and temperature vary significantly with altitude
- We're only looking at small variations from background values
 - so Boussinesq approximation is valid
- θ is temperature resulting from the same heat in the (dry) air, but at sea-level pressure
 - "adiabatically" moved down to sea-level
 - Ignoring small variations from background values
 - So $\boldsymbol{\theta}$ is more of a measure of heat, but in units of temperature

Buoyancy

 Again, just buoyancy force is just the variation in force due to gravity resulting from variation in density

$$f_{buoyancy} = \frac{\rho_{background} + \Delta \rho}{\rho_{background}} g$$

- Adding or subtracting a constant body force (e.g. g) is compensated by linear pressure gradient
- So we can cancel that part out, left with

$$f_{buoyancy} = \frac{\Delta \rho}{\rho_{background}} g$$

Calculating buoyancy

- Relative density changes due to thermal expansion
 - Also add in effect of water vapour in the air

 changes heat capacity and density, so a given potential temperature results in a different thermal expansion
- Presence of condensed water also increases density

$$g\left(\frac{\theta(1+0.61q_{v})}{\theta_{v0}}-q_{c}\right)$$

Condensation

- q_v and q_c advect around, and convert from one to the other (q_v+q_c conserved)
- Rate of condensation: $\frac{Dq_c}{Dt} = C$, $\frac{Dq_v}{Dt} = -C$
- Very simple model of condensation:
 - Water evaporates instantly up to the saturation point, any excess water condenses
 - Compute C to make sure this holds
- Critical value is saturation point $\ensuremath{\mathsf{q}_{vs}}$
 - Depends on pressure, absolute temperature, ...

Saturation point

• Empirical model:

$$q_{vs} = \frac{380.16}{p} \exp\left(\frac{17.67 T}{T + 243.5}\right)$$

• Background pressure p is

$$p = p_0 \left(1 - \frac{y\Gamma}{T_0} \right)^{\frac{\sigma}{\Gamma R_{dry}}}$$

- With pressure p₀≈101Kpa and temperature T₀≈300K (in Kelvins) at sea-level, Γ≈10K/km is the "lapse rate" or how fast temperature drops with altitude near Earth, and R_{dry}≈287J/(kg K) is the gas constant for dry air
- Temperature T is in Celsius...

Saturation point cont'd

 Temperature T_K in Kelvin (from absolute zero) is based on potential temperature:

$$T_{K} = \Pi \theta$$
$$= \left(\frac{p}{p_{0}}\right)^{\kappa} \theta = \left(\frac{p}{p_{0}}\right)^{\frac{R_{dry}}{c_{p}}} \theta \approx \left(\frac{p}{p_{0}}\right)^{0.286} \theta$$

- Π is the "Exner function", c_p≈ 1000J/(kg K) is specific heat of dry air at constant volume
- For saturation point, use ${\rm T_0}$ for θ
- Subtract 287K to get T in Celsius

Latent heat

- One last effect
 - As water condenses, it releases latent heat
 - · As water evaporates, it takes heat
 - (How your fridge works)
- So potential temperature is advected around, and changes according to rate of condensation:

$$\frac{D\theta}{Dt} = \frac{L}{c_p \Pi} C$$

• Here L≈2.5J/kg is latent heat

Putting it together

- Advect quantities around
- Add buoyancy force and others (vorticity confinement)
- Calculate condensation amount $C\Delta t = max(q_v-q_{vs}, -q_c)$
- Update $q_v,\,q_c$ and θ from ΔC
- Do pressure solve to enforce incompressibility

What next?

- Other phenomena
 - Geophysical flows (hurricanes?)
 - Freezing, melting, boiling, ...
 - Multiphysics
- More subtle physics effects
 - Avoid some of the simplifying assumptions
 - Higher accuracy numerical methods
- Fighting resolution
 - For 3D we scale like $O(n^3)$ at best, often worse
 - Adaptive methods, clever hacks
- Control, user-interface

Final notes

- Marking yes, I'm behind
- I will try to get higher quality software renderers up on the web (for the assignments) as soon as I can
- I'll post movies of the assignments on the web